

269(4): The z Component of the Hamiltonian
and its Elliptical Representation

From first principles:

$$L_z = m r^2 \dot{\phi} \sin^2 \theta \quad - (1)$$

and

$$\dot{\phi} = \frac{L_z}{m r^2 \sin^2 \theta} \quad - (2)$$

$$= \frac{L_1}{m r^2}$$

The relevant Hamiltonian is:

$$H_z = \frac{1}{2} m (\dot{r}^2 + r^2 \dot{\phi}^2) - \frac{k}{r} \quad - (3)$$

which gives the ellipse:

$$r = \frac{d_1}{1 + E_1 \cos \phi} \quad - (4)$$

where:

$$d_1 = \frac{L_1^2}{m k} = \frac{L_z^2}{m k \sin^4 \theta} \quad - (5)$$

and

$$E_1^2 = 1 + \frac{2 E L_1^2}{m k^2} = 1 + \frac{2 E L_z^2}{m k^2 \sin^4 \theta} \quad - (6)$$

2) O_L quantization:

$$L_z \psi = m_l \hbar \psi \quad - (7)$$

where $m_l = l, l-1, \dots, -l \quad - (8)$

and $\langle L_z^2 \rangle = m_l^2 \hbar^2 \quad - (8)$

In atomic hydrogen:

$$E = - \frac{m e^4}{32 \pi^2 \epsilon_0^2 \hbar^2 n^2} \quad - (9)$$

and $k = \frac{e^2}{4 \pi \epsilon_0} \quad - (10)$

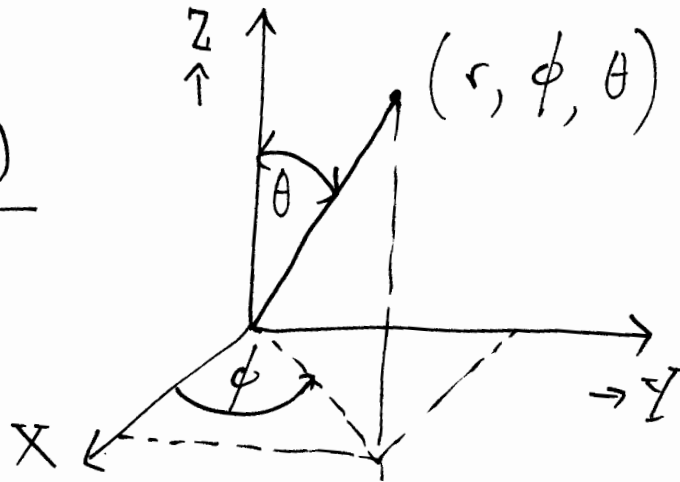
Suggested computer work

The function (4) can be plotted in a three dimensional spherical polar plot in which r is a function of θ and ϕ .

The complete function is:

$$r = \frac{L_z^2}{n k \sin^4 \theta} \left[1 + \left(1 + \frac{2 E L_z^2}{n k^2 \sin^4 \theta} \right)^{1/2} \cos \phi \right]^{-1} \quad - (7)$$

Fig. (1)



Here:

$$X = r \sin \theta \cos \phi \quad - (8)$$

$$Y = r \sin \theta \sin \phi \quad - (9)$$

$$Z = r \cos \theta \quad - (10)$$

$$\sin \phi = \frac{Y}{(X^2 + Y^2)^{1/2}} \quad - (11)$$

$$\cos \phi = \frac{X}{(X^2 + Y^2)^{1/2}} \quad - (12)$$

$$\cos \theta = \frac{Z}{(X^2 + Y^2 + Z^2)^{1/2}} \quad (13)$$

$$\sin^2 \theta = 1 - \cos^2 \theta \quad - (14)$$

Therefore r on the left hand side of eq. (7) can be expressed as a function of x, y and z or of θ and ϕ . On the classical level E and L_z can be chosen as parameters.

On the quantum level only the eigenvalues of L_z and L^2 operators are observable.

It is possible to extend this method to L_x and L_y and this will be done in the next note.
